



Flue gas analysis (orsat method)

The mixture of gases (like CO2, O2 & CO) coming out from the combustion chamber is called flue gas.

- The analysis of a flue gas would give an idea about the complete or incomplete combustion process.
- > The analysis of flue gas is carried out by using **Orsat's apparatus.**

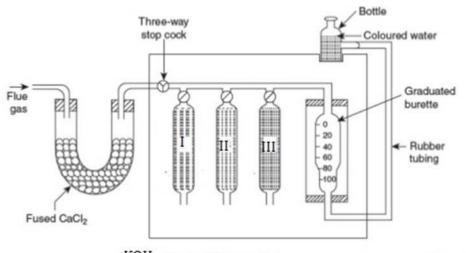
#### **Description of Orsat's apparatus**

- It consists of a horizontal tube.
- At one end of this tube, 'U' tube containing fused CaCl2 is connected through 3 way stop cock.
- > The other end of the tube is connected with a graduated burette.
- The burette is surrounded by a water jacket to keep the temperature of the gas constant.
- The lower end of the burette is connected to a water reservoir by means of a rubber tube.
- The level of water in the burette can be raised or lowered by raising or lowering the reservoir.
- The horizontal tube is also connected with three different absorption bulbs 1, 2 and 3 for absorbing CO2, O2 and CO.

Bulb 1 contains KOH and it absorbs CO2 only.

Bulb 2 contains alkaline pyrogallol and it absorbs CO2 and O2.

Bulb 3 contains ammoniacal cuprous chloride and it absorbs CO2, O2 and CO.



KOH alkaline Pyrogallol ammonia cal cuprous Chloride

#### Working



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- The three way stop cock is opened to the atmosphere and the burette is completely filled with water and air is sent out.
- The burette is filled with flue gas to 100 cc by raising or lowering the reservoir. Now the 3- way stop cock is closed.

## 1. Absorption of CO2

- The bulb 1 is opened and all the gas is passed into bulb1 by raising the level of water in the burette.
- The gas enters into bulb1 where CO2 is absorbed by KOH. The gas is again sent to the burette.
- > The process is repeated several times to ensure complete absorption of CO2.
- The decrease in volume of the flue gas = the volume of CO2 in 100cc of the flue gas.

### 2. Absorption of O2

- Bulb 1 is closed and bulb 2 is opened.
- The gas is again sent into bulb 2 where O2 in the flue gas is absorbed by alkaline pyrogallol.
- The decrease in volume of the flue gas = the volume of O2.

## 3. Absorption of CO

- Bulb 2 is closed and bulb 3 is opened.
- The remaining gas is sent into bulb 3, where CO is absorbed by ammoniacal cuprous chloride.
- > The decrease in volume of flue gas = the volume of CO.
- The remaining gas in the burette after the absorption of CO2, O2 and CO is taken as nitrogen.

#### Significance

- i) It gives an idea about the complete or incomplete combustion.
- ii) If the flue gas contains considerable amount of CO, it indicates incomplete combustion and short supply of O2.
- iii) If the flue gas contain considerable amount of O<sub>2</sub>, it indicates complete combustion and excess supply of O<sub>2</sub>.