

### Problem on Indicated Power (IP)

A single-cylinder, four-stroke engine has a cylinder bore of 0.1 meters and a stroke length of 0.12 meters. The engine operates at 3000 RPM with an average effective pressure of 800 kPa. Calculate the indicated power of the engine.

Given:

- Cylinder bore ( $D$ ) = 0.1 m
- Stroke length ( $L$ ) = 0.12 m
- Engine speed ( $N$ ) = 3000 RPM
- Mean effective pressure ( $P_{mean}$ ) = 800 kPa = 800,000 Pa

First, calculate the cross-sectional area of the cylinder:

$$A = \frac{\pi D^2}{4} = \frac{\pi(0.1)^2}{4} = 0.00785 \text{ m}^2$$

Since it is a four-stroke engine, it completes one power stroke every two revolutions:

$$\text{Power strokes per minute} = \frac{N}{2} = \frac{3000}{2} = 1500 \text{ strokes/min}$$

The indicated power (IP) is given by:

$$IP = \frac{P_m \downarrow \cdot L \cdot A \cdot N \cdot K}{60}$$

For a single-cylinder engine ( $K = 1$ ):

$$IP = \frac{800,000 \times 0.12 \times 0.00785 \times 1500 \times 1}{60}$$

$$IP = 1884 \text{ W} = 1.884 \text{ kW}$$

### Problem on Thermal Power

An internal combustion engine consumes 0.02 kg/s of fuel with a calorific value of 42,000 kJ/kg. Calculate the thermal power generated by the fuel.

Given:

- Fuel consumption rate ( $\dot{m}_f$ ) = 0.02 kg/s
- Calorific value ( $CV$ ) = 42,000 kJ/kg = 42,000,000 J/kg

The thermal power ( $Q_{th}$ ) is given by:

$$Q_{th} = \dot{m}_f \cdot CV$$

$$Q_{th} = 0.02 \times 42,000,000 = 840,000 \text{ W} = 840 \text{ kW}$$

### Problem on Indicated Thermal Efficiency

An engine has an indicated power of 75 kW. It consumes fuel at a rate of 0.015 kg/s, and the fuel has a calorific value of 44,000 kJ/kg. Calculate the indicated thermal efficiency of the engine.

Given:

- Indicated power ( $IP$ ) = 75 kW
- Fuel consumption rate ( $\dot{m}_f$ ) = 0.015 kg/s
- Calorific value ( $CV$ ) = 44,000 kJ/kg = 44,000,000 J/kg

First, calculate the thermal power ( $Q_{th}$ ):

$$Q_{th} = \dot{m}_f \cdot CV$$

$$Q_{th} = 0.015 \times 44,000,000 = 660,000 \text{ W} = 660 \text{ kW}$$

Now, calculate the indicated thermal efficiency ( $\eta_{ith}$ ):

$$\eta_{ith} = \frac{IP}{Q_{th}} = \frac{75}{660} = 0.1136 \text{ or } 11.36\%$$