

# Adsorption,column,partition,affinity, ion exchange,size exclusion

various **separation techniques** are applied to isolate, purify, and concentrate specific components from complex food matrices. Each technique relies on different physical or chemical properties of the substances to achieve the separation. Here's an overview of the mentioned techniques:

### 1. Adsorption

- **Principle**: Adsorption separation relies on the adherence of molecules to a solid surface (adsorbent). Different molecules have different affinities to the adsorbent, leading to selective adsorption.
- **Application in Food**: Adsorption is used for decolorizing, removing off-flavors, or purifying proteins and enzymes in food processing. Activated charcoal or silica can be used to remove impurities.
- **Example**: Removing unwanted flavors from vegetable oils using adsorbents like bentonite.

# 2. Column Chromatography

- **Principle**: This technique separates compounds based on their interactions with a stationary phase packed in a column and a mobile phase that passes through the column.
- Application in Food: It is used for isolating vitamins, sugars, and pigments, and for purifying bioactive compounds like antioxidants and flavors.
- **Example**: Separation of proteins and peptides from food extracts using gel or ion-exchange chromatography.



www.snsgroups.com

### **3.** Partition Chromatography

- **Principle**: Separation is based on the difference in solubility of components between two immiscible phases, usually one stationary liquid phase and one mobile liquid phase.
- **Application in Food**: It helps in the separation of lipid-soluble compounds from aqueous mixtures.
- **Example**: Extracting essential oils or flavors from complex food matrices.





# Adsorption and Partition Column Chromatography



# 4. Affinity Chromatography

- **Principle**: This technique relies on the specific interaction between a molecule (ligand) attached to the stationary phase and a target molecule in the sample. Only the target molecule binds, while others are washed away.
- **Application in Food**: Used for purifying proteins, enzymes, or other bioactive compounds that have specific binding sites.
- **Example**: Isolating enzymes or antibodies from food extracts.

The image	The image	The image
part with	part with	part with
relationship	relationship	relationship
ID rid8 was	ID rid8 was	ID rid8 was
not found in	not found in	not found in
the file.	the file.	the file.

#### UNIT 3 CHROMATOGRAPHIC AND SEPARATION TECHNIQUES





### 5. Ion Exchange Chromatography

- **Principle**: Molecules are separated based on their charge. An ion-exchange resin (cationic or anionic) binds molecules of opposite charge, allowing for the separation of charged molecules.
- **Application in Food**: It's used to purify proteins, desalinate solutions, or remove unwanted ions (e.g., in water softening).
- **Example**: Purification of amino acids and proteins in dairy products.

### 6. Size Exclusion Chromatography (SEC)

- **Principle**: This technique separates molecules based on their size by passing them through a column filled with porous beads. Larger molecules are excluded from entering the pores and elute first, while smaller molecules enter the pores and elute later.
- Application in Food: Size exclusion is commonly used to separate proteins, polysaccharides, and other macromolecules.
- **Example**: Separation of lactose from dairy products or separating protein fractions based on molecular size.