



# **SNS COLLEGE OF TECHNOLOGY**

(An Autonomous Institution)

COIMBATORE-35

**Accredited by NBA-AICTE and Accredited by NAAC – UGC with A+ Grade**

**Approved by AICTE, New Delhi & Affiliated to Anna University, Chennai**

## **DEPARTMENT OF ELECTRICAL AND ELECTRONICS ENGINEERING**



**COURSE NAME: 19EEO305 /Renewable Energy Generation Technology**

**IV YEAR / VII SEMESTER**

**UNIT 4- BIOMASS AND HYDRO ENERGY**

**Topic 1 – Biomass direct combustion**



# SUCCESSFUL STUDENT

Positive  
Attitude

Professionally  
Groomed

Socially  
Interactive

Technically  
Skillful



# Solid fuels

## Fossil fuels:

- Coal (moisture, volatiles, fixed carbon, ash)  
( $\text{CH}_{0.8}$ )
- Coke (devolatilized coal or petroleum)

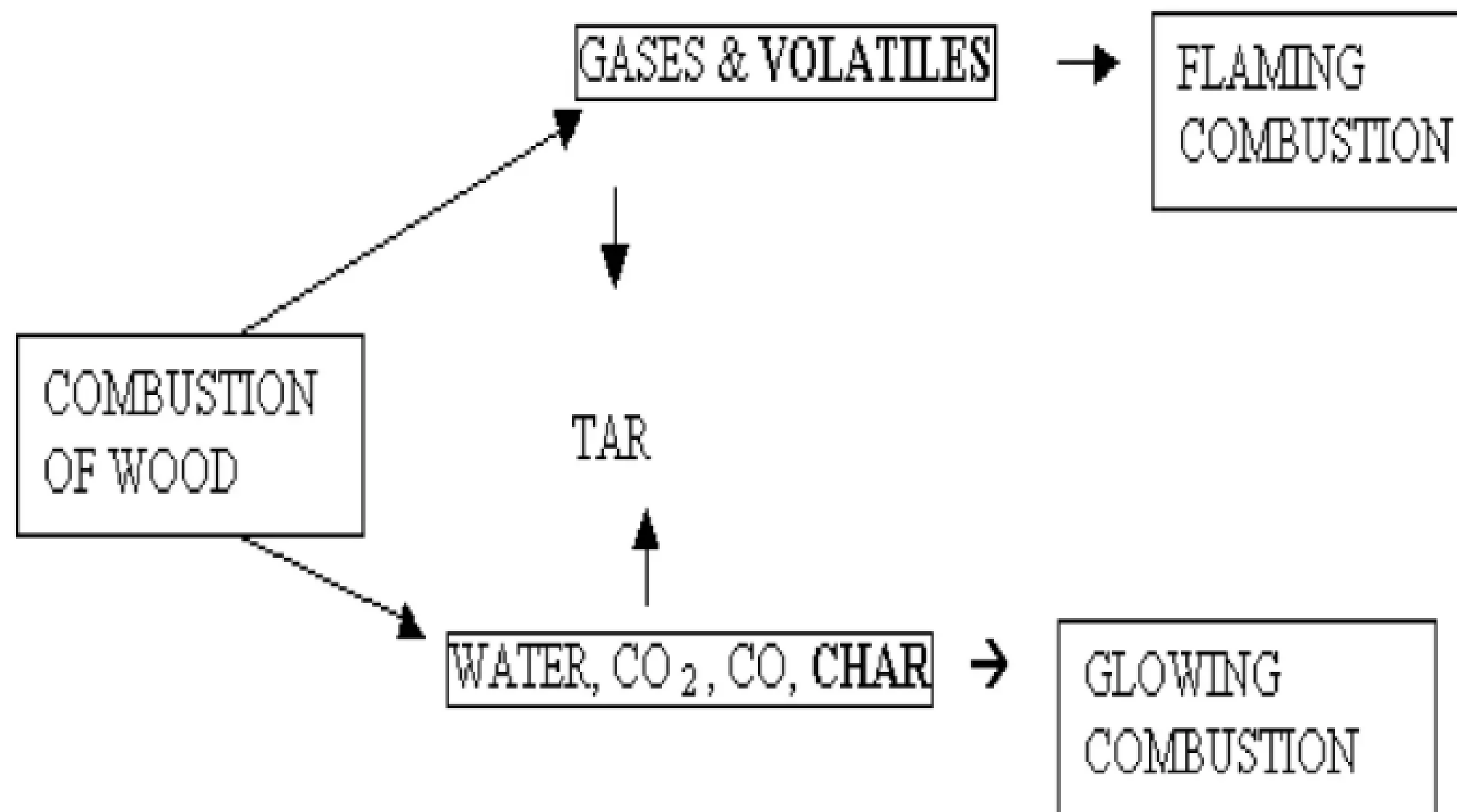
## Biofuels:

- Wood (moisture, volatiles, fixed carbon, ash)
- Charcoal (devolatilized wood)
- Key difference among fuels: the quantity of  $\text{CO}_2$  formed per unit of energy released. Natural gas releases  $\sim 42\%$  less  $\text{CO}_2$  than coal



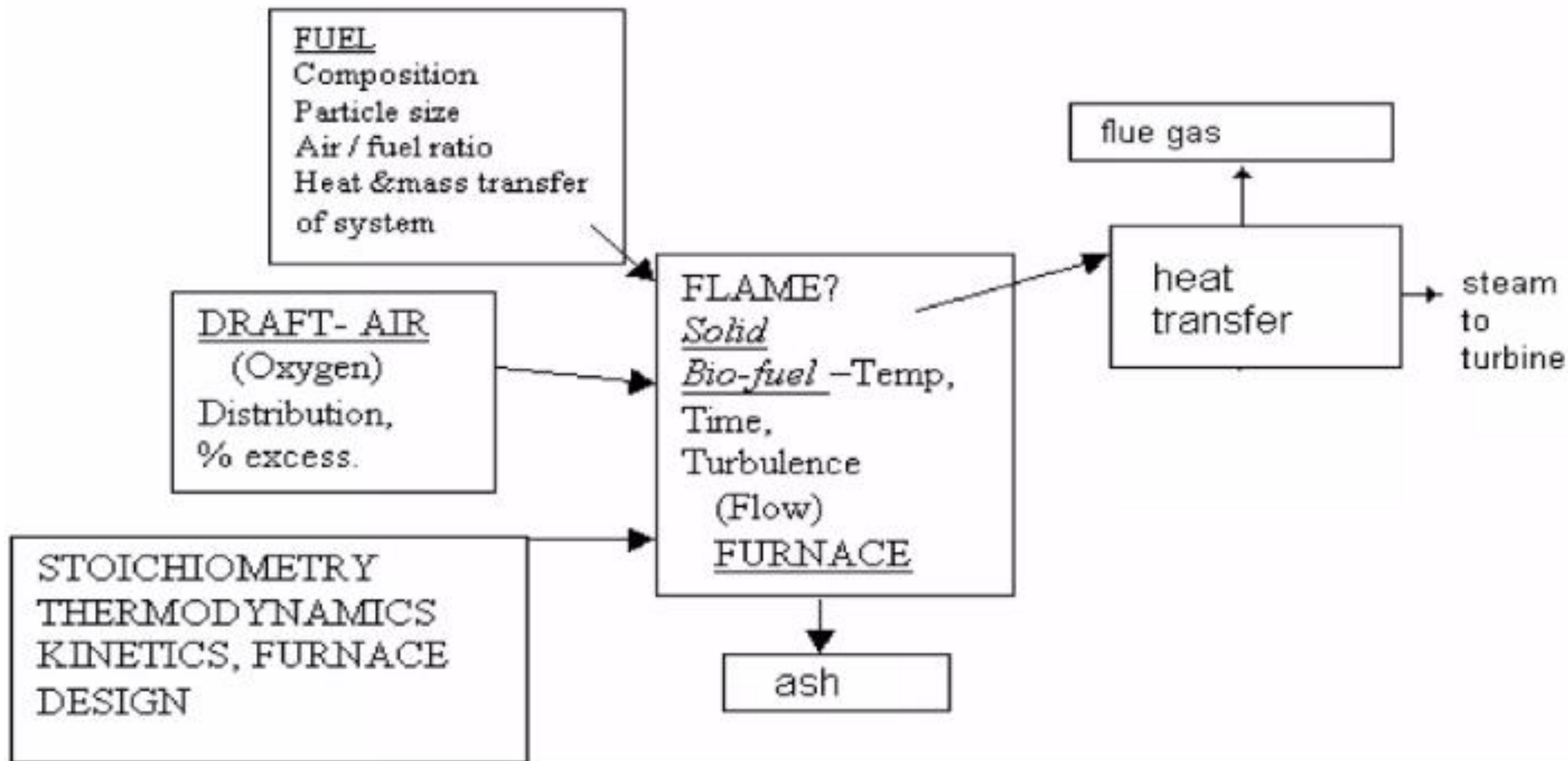
Chemical equilibrium constant are calculated for following 6 equations. Eq. No.3 & 4 is called as Water Gas Reaction. Eq. No.5 is called Methane Formation Reaction. Eq. No.6 is called as Shift Reaction

| Chemical Equations        | $\Delta H_0$ | $\Delta S_0$ | Equation No. |
|---------------------------|--------------|--------------|--------------|
| $C + O_2 = CO_2$          | -94,200      | 2.06         | 1            |
| $2C + O_2 = 2CO$          | -53,300      | 45.54        | 2            |
| $C + H_2O = CO + H_2$     | +31,230      | 33.41        | 3            |
| $C + 2H_2O = CO_2 + H_2$  | +21,560      | -7.89        | 4            |
| $CO + 3H_2 = CH_4 + H_2O$ | -49,300      | 6.51         | 5            |
| $CO + H_2O = CO_2 + H_2$  | -9,670       | -10.07       | 6            |



**Fig: General pathways for pyrolysis and combustion of Wood**

Combustion: A chemical process \_ Oxidation of reduced forms of carbon and hydrogen by free radical processes. Chemical properties of the bio-fuels determine the higher heating value of the fuel and the pathways of combustion.



COMBUSTION OF BIOFUEL IN BOILER TO GENERATE STEAM



## COMPARISON OF COAL AND WOOD AS FUEL FOR COMBUSTION:

### COAL

- Solid fuel, high ash content,
- used for Raising HP steam,
- Power production with Rankine cycle
- Gas Turbine cycles, Brayton cycle
- Can be used for producing process steam for direct heating
- Large scale availability near mines and ports
- Assured Technology for handling, storage and Processing well established
- Sulfur content and ash content are problems

### WOOD

- Solid fuel, less ash, more volatile, reactive,
- used for Raising HP steam,
- Power production with Rankine cycle,
- Gas Turbine cycles more difficult
- Can be used for producing process steam for direct heating
- Assured availability is only on small scale— Variable
- Large scale processing, storage and energy conversion technology not established in India
- Moisture content, low bulk density, Location specific availability are problems



# ASSESSMENT



publicdomainvectors.org







# REFERENCE



## Reference Book:

1. S.P. Sukhatme, 'Solar Energy', Tata McGraw Hill Publishing Company Ltd., New Delhi, 1997. (UNIT II)
2. G.N. Tiwari, 'Solar Energy – Fundamentals Design, Modelling and applications', Narosa Publishing House, New Delhi, 2002. (UNIT II)
3. S.M. Muyeen," Wind Energy Conversion Systems: Technology and Trends", Springer 2012. [UNIT III]

## Text Book:

1. G.D. Rai, 'Non Conventional Energy Sources', Khanna Publishers, New Delhi, 2006. (UNIT I - V)
2. D.P.Kothari, K.C.Singal and Rakesh Ranjan,"Renewable energy sources and Emerging Technologies", PHI Pvt. Ltd., 2009. (UNIT I-V)



# THANK YOU!!

