

#### IMPRESSED CURRENT CATHODIC PROTECTION METHOD



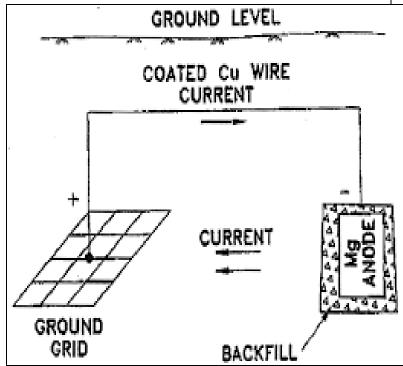
#### **PRINCIPLE**

An impressed current is applied in the opposite direction of the corrosion current to nullify it & corroding metal is converted from

anode to cathode.

### **Construction**

- > -Ve terminal : object to be protected
- >+Ve terminal : an inert anode.
- ➤ Inert anode: graphite or platinized Ti
- > The anode is buried in a back fill



➤ (Back fill: a mixture of gypsum, coke, breeze and sodium



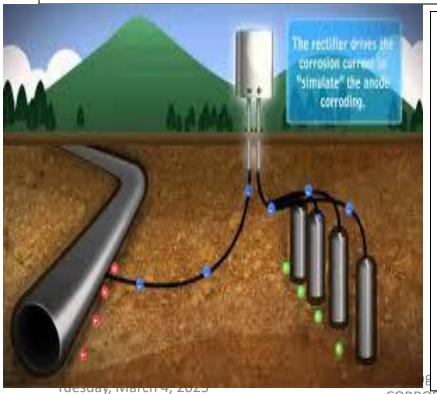
#### **WORKING**

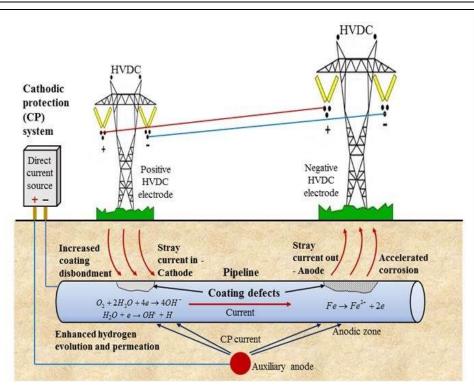


**The** back fill provides good electrical contact to anode.

The current from the battery is impressed on the metallic structure to be protected which acts as the cathode.

# **Applications**





CORROSION & ITS CONTROL





### Advantages of impressed current protection method

- oLarger driving voltage.
- oLarger flexibility control.
- oIt is applicable to large objects.
- OUncoated parts can also be protected.

## Limitations of impressed current protection method

1. Maintenance and installation cost are very high.